

Brief Articles

Synthesis and Structure–Activity Relationship of Fluoro Analogues of 8-{2-[4-(4-Methoxyphenyl)piperazin-1-yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione as Selective α_{1D} -Adrenergic Receptor Antagonists

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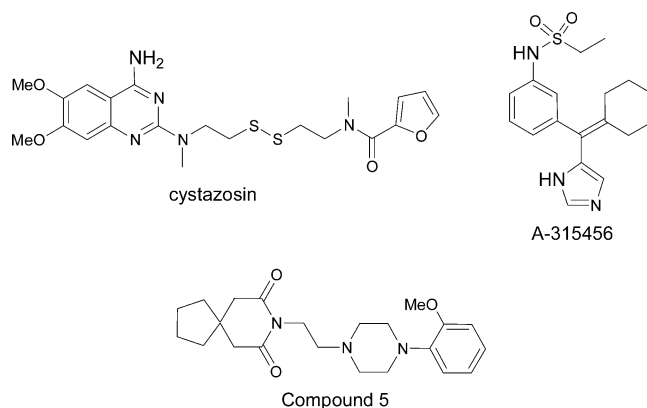
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We have discovered high-affinity antagonists (exemplified by **11** and **12**) that are the most selective for α_{1D} -adrenergic receptors (α_{1D} -AR) reported to date. In cloned receptor assay systems, **12** displays at least 95-fold selectivity for the α_{1D} -AR over all other G-protein-coupled receptors tested, and the subtype selectivity of **11** was confirmed in pharmacologically defined isolated tissue preparations.

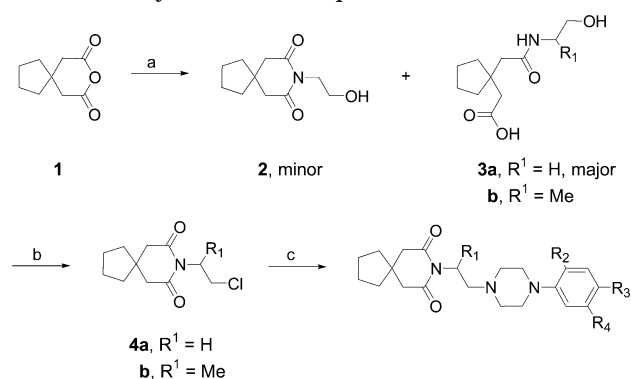
α -Adrenergic receptors (α -ARs) modulate intercellular biochemical processes in response to changes in extracellular concentrations of the neurotransmitter norepinephrine and the circulating hormone epinephrine, leading to widespread physiological actions that make them attractive targets for drug discovery.¹ Antagonists that are highly selective for the α_{1A} -AR are well-known,^{1c} and recent reports have described compounds with modest selectivity for the α_{1B} -AR.² Saussy et al. have reported that **5** (Chart 1)³ is selective for the α_{1D} -AR,⁴ and this has been confirmed in our assay systems (Table 1). However, **5** ($pK_i = 8.8$ at α_{1D}) also has comparable or higher affinity for several other G-protein-coupled receptors (GPCRs),⁴ including the serotonin 5-HT_{1A} receptor and the dopamine D₂ and D₃ receptors (Table 1). Cystazosin⁵ has also been reported to be selective for the α_{1D} -AR and devoid of cross-reactivity to 5-HT_{1A} and dopamine receptors. Its subtype selectivity, however, is only about 10-fold. Recently, A-315456 has been reported⁶ to be selective for the α_{1D} -AR with low cross reactivity to 5-HT_{1A} and D₂. Also, recently a report by a group from Recordati on analogues of **5** with low cross-reactivity has appeared.⁷ The report did not mention whether the Recordati compounds had significant cross-reactivity to dopamine receptors, a known cross-reactivity of **5**.⁶

We describe herein the synthesis and SAR of fluoro analogues of **5** and novel trifluoro analogues that show decreased affinity for 5-HT_{1A}, D₂, and D₃ receptors while maintaining high affinity and subtype selectivity for the α_{1D} -AR.^{8,9}

Chart 1



Scheme 1. Synthesis of Compounds 6–13^a



^a Reagents and conditions: (a) 2-ethanolamine or 2-aminopropanol; (b) SOCl₂, 45% from **1**; (c) substituted *N*-phenylpiperazine, 22–50%.

Synthesis

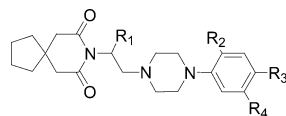
Compounds **6–13** were synthesized as outlined in Scheme 1. 3,3-Tetramethyleneglutaric anhydride (**1**) was allowed to react with ethanolamine, giving a mixture of imide **2** and amide **3a**. The crude mixture of imide **2** and amide **3a** was treated with thionyl chloride,

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Table 1. Binding Affinities at Cloned α -Adrenoceptors, 5-HT_{1A}, and Dopamine Receptors

| compd | R1 | R2 | R3 | R4 | K_i (nM) ^a | | | | | |
|-----------|-----------------|-----|----|----|-------------------------|-----------------|-----------------|--------------------|-----------------|-----------------|
| | | | | | α_{1d} | α_{1b} | α_{1a} | 5-HT _{1A} | D ₂ | D ₃ |
| 5 | H | OMe | H | H | 1.6 | 191 | 290 | 0.46 | 14 | 3.1 |
| 6 | H | F | H | H | 0.83 | 67 | 630 | 7.5 | 140 | 5.2 |
| 7 | H | H | F | H | 14 | 255 | 620 | 54 | 2200 | 91 |
| 8 | H | F | F | H | 5.6 | 1100 | 3800 | 24 | NT ^b | NT ^b |
| 9 | H | H | F | F | 7.9 | 1400 | 4100 | 55 | 2200 | 160 |
| 10 | H | F | H | F | 0.95 | 54 | 1000 | 25 | 120 | 8.1 |
| 11 | H | F | F | F | 1.9 | 380 | 4400 | 360 | 700 | 23 |
| 12 | (<i>R</i>)-Me | F | F | F | 1.3 | 165 | 14000 | 300 | 580 | 120 |
| 13 | (<i>S</i>)-Me | F | F | F | >124 | NT ^b | NT ^b | NT ^b | NT ^b | NT ^b |

^a α_{1d} , α_{1b} , α_{1a} , 5-HT_{1A}, and D₂ are cloned human receptors. D₃ is a cloned rat receptor. K_i determinations are an average of two or more (four to six for key compounds) independent determinations. The margin of error is within 5% of the mean for all data shown. ^b NT: not tested.

resulting in condensation of the amide to the imide and substitution of the hydroxyl group with chloride. The resulting intermediates **4** were treated with the appropriately substituted *N*-arylpiperazines, giving the desired compounds **6–13**. The *N*-arylpiperazines were commercially available or synthesized by a previously described procedure.¹⁰ For **12** (and its enantiomer **13**), tetramethyleneglutaric anhydride (**1**) was allowed to react with optically pure (*R*)-2-aminopropanol (or (*S*)-2-aminopropanol), giving imide **3b** that was converted to **4b** by thionyl chloride treatment.

Pharmacology

Radioligand binding experiments were performed on membranes prepared from cells transiently transfected with DNA for the cloned human α -AR (α_{1a} , α_{1b} , and α_{1d}) and 5-HT_{1A}, as described previously.¹¹ Membranes for dopamine human D₂ and rat D₃ receptors were purchased from New England Nuclear Corporation. The binding affinities (K_i) were determined by displacement of the following radioligands: [³H]prazosin (0.3 nM, α_1 -ARs), [³H]-8-hydroxy-2-(di-*n*-propylamino)tetralin (8-OH-DPAT, 0.3 nM, 5-HT_{1A}), or [³H]spiperone (1 nM, D₂, D₃). The results are presented in Table 1.

The K_B of **11** at native α_1 -ARs was determined by measuring antagonism of phenylephrine-evoked contractions in three pharmacologically defined isolated rat tissue preparations: vas deferens (α_{1A}), spleen (α_{1B}), and thoracic aorta (α_{1D}). In each of the preparations **11** behaved as a competitive antagonist. The K_B determined for each tissue (α_{1A} , 5.4; α_{1B} , 6.7; α_{1D} , 8.8) correlates well with the corresponding K_i derived from binding experiments.

Structure–Activity Relationships

We hypothesized that the selectivity of **5** for the α_{1d} -AR over the α_{1a} - and α_{1b} -ARs could be largely attributed to the imide moiety because many examples were known of 1-substituted 4-(methoxyphenyl)piperazines that were not α_{1d} -selective.^{1c} Therefore, we focused our initial studies on modification of the piperazine moiety.

It was found that the methoxyl group of **5** could be replaced with fluorine (**6**), resulting in decreased affinity

for the 5-HT_{1A} and D₂ receptors (16- and 13-fold, respectively) but essentially unchanged affinity for the α_{1d} and D₃ receptors (Table 1). The 4-fluoro-substituted **7** displays a 18-fold decrease in α_{1d} -AR affinity and a decrease in the selectivity for α_{1d} versus 5-HT_{1A} compared to the 2-fluoro-substituted **6**. For difluoro substitution, the substitution pattern of the fluorine atoms on the phenyl ring is critical for maximizing the affinity and selectivity for the α_{1d} -AR. For instance, it was found that the affinity for α_{1d} for 2,5-difluoro-substituted **10** is equivalent to the 2-fluoro-substituted **6** while affinity for the 5-HT_{1A} and α_{1a} receptors is decreased. Compared to 2,5-difluoro substitution (**10**), the binding affinity and selectivity for the α_{1d} -AR decreased with 2,4-difluoro substitution (**8**) or 3,4-difluoro substitution (**9**). The trifluorophenyl-substituted **11** exhibits α_{1d} -AR affinity approximately equal to that of **5** while displaying significantly decreased affinity for α_{1a} , 5-HT_{1A}, and D₂ receptors (Table 1). The affinity of **11** for the D₃ receptor was reduced relative to **5** (6-fold) but was still significantly high ($K_i = 23$ nM).

Placement of a methyl group on the linker (**12**) resulted in decreased binding affinity at the D₃ receptor ($K_i = 123$ nM) while maintaining high affinity for the α_{1d} -AR ($K_i = 1.3$ nM) and greater than 100-fold selectivity over other GPCRs including α_{1a} , α_{1b} , α_{2c} , α_{2a} , α_{2b} , D₂, and 5-HT_{1A}. The (*R*)-configuration of the methyl group is important because the (*S*)-isomer was found to have significantly lower binding affinity for the α_{1d} -AR ($K_i \geq 124$ nM).¹²

In conclusion, we have discovered high-affinity antagonists (exemplified by **11** and **12**) that are the most selective for the α_{1d} -AR reported to date. In cloned receptor assay systems, **12** displays at least 95-fold selectivity for the α_{1d} -AR over all other GPCRs tested, and the subtype selectivity of **11** was confirmed in pharmacologically defined isolated tissue preparations. These compounds display the highest selectivity for the α_{1d} -AR thus far reported and should prove useful for further functional characterization of α_1 -ARs in *in vivo* models. The information gained through such studies will aid in the understanding of the physiological importance of the individual α_1 -AR subtypes and potentially lead to the discovery of therapies that benefit from selective modulation of α_1 -ARs.

Experimental Section

General Methods. Substituted *N*-phenylpiperazines were synthesized according to the procedure described by Martin.¹⁰ The syntheses of **4** were carried out as described previously by Y. H. Wu.³ (*R*)-2-Aminopropanol and (*S*)-2-aminopropanol were purchased from Aldrich and both are listed as 97% ee. The actual ee purity for the batch of (*R*)-2-aminopropanol used in the syntheses described herein was 99.9% (GLC) as communicated to us by Aldrich. Enantiomeric purities of **12** and **13** were determined by chiral HPLC, using a Chiralcel OD, 0.46 cm \times 25 cm column (Daicel Chemical Industries, LTD), 1 mL/min (5% EtOH/95% hexane with 0.1% TEA). ¹H and ¹³C NMR spectra were obtained at 300 and 75 MHz, respectively, with CDCl₃ as solvent and referenced to TMS as an internal standard. Coupling constants (*J*) are reported in Hz.

8-[(1*R*)-2-Chloro-1-methylethyl]-8-azaspiro[4.5]decane-7,9-dione (4b**).** A mixture of 3,3-tetramethyleneglutaric anhydride (1.12 g, 6.66 mmol) and (*R*)-(-)-2-amino-1-propanol (Aldrich, 1.00 g, 13.3 mmol) in pyridine (15 mL) was heated at reflux for 3 h. The solvent was removed, and the residue was partitioned between 1 N HCl (10 mL) and ethyl acetate

(10 mL). The aqueous layer was extracted with ethyl acetate (2 × 10 mL). The combined ethyl acetate fractions were dried over sodium sulfate and then the solvent was removed, leaving a clear oil (1.92 g). A portion of this oil (0.70 g) in benzene (9 mL) and pyridine (0.40 mL) was cooled to 0 °C. Thionyl chloride (0.40 mL) was added dropwise to the mixture, and then the solution was heated at 60 °C for 90 min. The solution was cooled to room temperature, and water (10 mL) was added. The layers were separated, and the aqueous layer was extracted with ethyl acetate (2 × 10 mL). The solvent was removed from the combined organic fractions, and the residue was purified by flash chromatography over silica gel, eluting with hexane/ethyl acetate (3:1). The $R_f = 0.3$ fraction was concentrated, giving the title compound as a pale-yellow oil (294 mg). $^1\text{H NMR}$ δ 5.09–5.04 (m, 1H), 4.17 (t, 1H, $J = 10.5$), 3.66 (dd, 1H, $J = 11.1, 5.7$), 2.61 (s, 4H), 1.76–1.69 (m, 4H), 1.56–1.51 (m, 4H), 1.40 (d, 3H, $J = 6.9$). The (*S*)-enantiomer was synthesized in an identical manner.

8-{2-[4-(2-Fluorophenyl)piperazin-1yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (6). A mixture of 1-(2-fluorophenyl)piperazine (100 mg, 0.56 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (100 mg, 0.44 mmol) was heated with stirring at 160 °C for 5 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.3$ was removed and rinsed with chloroform/methanol (4:1). The solvent was removed, giving the title compound as pale-yellow oil (81.7 mg, 0.22 mmol, 50%). $^1\text{H NMR}$ δ 7.04 (td, 1H, $J = 9.3, 1.5$), 7.01 (t, 1H, $J = 6.9$), 6.97–6.87 (m, 2H), 3.96 (t, 2H, $J = 6.6$), 3.05 (t, 4H, $J = 4.8$), 2.67 (t, 4H, $J = 4.7$), 2.59 (s, 4H), 2.54 (t, 2H, $J = 6.6$), 1.73–1.68 (m, 4H), 1.55–1.50 (m, 4H); $^{13}\text{C NMR}$ δ 172.8, 156.3 (d, $J = 244.4$), 140.8 (d, $J = 8.5$), 125.0 (d, $J = 3.4$), 122.8 (d, $J = 8.0$), 119.5 (d, $J = 2.9$), 116.7 (d, $J = 20.7$), 56.1, 53.9 (2C), 51.2 (d, 2C, $J = 3.1$), 45.5 (2C), 40.2, 38.1 (2C), 37.1, 24.8 (2C); ESI-MS m/z 374 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (mp 212–214 °C).

8-{2-[4-(4-Fluorophenyl)piperazin-1-yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (7). A mixture of 1-(4-fluorophenyl)piperazine (100 mg, 0.56 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (100 mg, 0.44 mmol) was heated with stirring at 160 °C for 7 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.3$ was removed and rinsed with chloroform/methanol (4:1). The solvent was removed, giving the title compound as pale-yellow oil (77.3 mg, 0.21 mmol, 48%). $^1\text{H NMR}$ δ 6.95 (dd, 2H, $J = 9.3, 8.1$), 6.85 (dd, 2H, $J = 9.3, 4.8$), 3.96 (t, 2H, $J = 6.6$), 3.05 (t, 4H, $J = 5.0$), 2.65 (t, 4H, $J = 4.8$), 2.59 (s, 4H), 2.54 (t, 2H, $J = 6.5$), 1.72–1.67 (m, 4H), 1.54–1.49 (m, 4H); $^{13}\text{C NMR}$ δ 172.9, 157.6 (d, $J = 237.1$), 148.6, 118.2 (d, 2C, $J = 7.6$), 116.0 (d, 2C, $J = 21.9$), 56.0, 53.8 (2C), 50.8 (2C), 45.5 (2C), 40.2, 38.1 (2C), 37.1, 24.8 (2C); ESI-MS m/z 374 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (mp 223–224 °C).

8-{2-[4-(3-Fluorophenyl)piperazin-1yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (8). A mixture of 1-(4-fluorophenyl)piperazine (100 mg, 0.56 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (100 mg, 0.44 mmol) was heated with stirring at 160 °C for 7 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.3$ was removed and rinsed with chloroform/methanol (4:1). The solvent was removed, giving the title compound as a pale-yellow oil (63.0 mg, 0.17 mmol, 39%). $^1\text{H NMR}$ δ 7.17 (q, 1H, $J = 8.1$), 6.65 (dd, 1H, $J = 8.4, 2.1$), 6.57 (dt, 1H, $J = 12.3, 2.3$), 6.51 (td, 1H, $J = 8.1, 2.1$), 3.96 (t, 2H, $J = 6.5$), 3.13 (t, 4H, $J = 5.0$), 2.63 (t, 4H, $J = 5.0$), 2.59 (s, 4H), 2.53 (t, 2H, $J = 6.6$), 1.74–1.64 (m, 4H), 1.52–1.48 (m, 4H); $^{13}\text{C NMR}$ δ 172.9, 164.4 (d, $J = 241.6$), 153.6 (d, $J = 9.2$), 130.6 (d, $J = 9.8$), 111.6 (d, $J = 2.2$), 106.2 (d, $J = 21.4$), 103.1 (d, $J = 24.8$), 56.0, 53.6 (2C), 49.3 (2C),

45.5 (2C), 40.2, 38.1 (2C), 37.0, 24.8 (2C); ESI-MS m/z 374 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (mp 238–229.5 °C).

8-{2-[4-(3,4-Difluorophenyl)piperazin-1yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (9). A mixture of 1-(3,4-difluorophenyl)piperazine (100 mg, 0.51 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (100 mg, 0.44 mmol) was heated with stirring at 160 °C for 7 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.3$ was removed and rinsed with chloroform/methanol (4:1). The solvent was removed, giving the title compound as a pale-yellow oil (69.8 mg, 0.18 mmol, 41%). $^1\text{H NMR}$ δ 7.17 (q, 1H, $J = 9.3$), 6.65 (ddd, 1H, $J = 13.5, 6.9, 3.0$), 6.59–6.54 (m, 1H), 3.97 (t, 2H, $J = 6.5$), 3.07 (t, 4H, $J = 4.8$), 2.68 (t, 4H, $J = 5.0$), 2.60 (s, 4H), 2.58 (t, 2H, $J = 6.3$), 1.72–1.68 (m, 4H), 1.54–1.50 (m, 4H); ESI-MS m/z 392 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving white flakes (mp 227–228 °C).

8-{2-[4-(2,5-Difluorophenyl)piperazin-1yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (10). A mixture of 1-(2,5-difluorophenyl)piperazine (100 mg, 0.51 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (100 mg, 0.44 mmol) was heated with stirring at 160 °C for 5 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.7$ was removed and rinsed with chloroform/methanol (4:1). The solvent was removed, giving the title compound as a pale-yellow oil. $^1\text{H NMR}$ δ 6.99–6.89 (11-line m, 1H), 6.65–6.52 (m, 2H), 3.95 (t, 2H, $J = 6.5$), 3.03 (t, 4H, $J = 4.7$), 2.66 (t, 4H, $J = 4.7$), 2.60 (s, 4H), 2.54 (t, 2H, $J = 6.6$), 1.74–1.69 (m, 4H), 1.55–1.51 (m, 4H); ESI-MS m/z 392 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (64.9 mg, 0.15 mmol, 35%) (mp 237–239 °C).

8-{2-[4-(2,4,5-Trifluorophenyl)piperazin-1yl]ethyl}-8-azaspiro[4.5]decane-7,9-dione (11). A mixture of 1-(2,4,5-trifluorophenyl)piperazine (0.94 g, 4.35 mmol) and 8-(2-chloroethyl)-8-azaspiro[4.5]decane-7,9-dione (1.00 g, 4.35 mmol) was heated with stirring at 160 °C for 7 h. The residue was partitioned between ethyl acetate (40 mL) and saturated aqueous sodium carbonate (40 mL). The aqueous layer was extracted with ethyl acetate (2 × 40 mL), and the combined ethyl acetate fractions dried over sodium sulfate. The solvent was removed, and the residue was purified by flash chromatography over silica gel, eluting with a gradient of hexane to hexane/ethyl acetate (1:1). The solvent was removed from the desired product [$R_f = 0.7$, hexane/ethyl acetate (1:1)], leaving a pale-tan oil that slowly solidified (0.652 g, 1.60 mmol, 37%, mp 230–234 °C). $^1\text{H NMR}$ δ 6.89 (ddd, 1H, $J = 11.7, 10.2, 7.5$), 6.74 (dt, 1H, $J = 12.0, 8.1$), 3.95 (t, 2H, $J = 6.6$), 2.97 (t, 4H, $J = 4.7$), 2.65 (t, 4H, $J = 4.7$), 2.60 (s, 4H), 2.54 (t, 2H, $J = 6.5$), 1.74–1.70 (m, 4H), 1.55–1.51 (m, 4H); $^{13}\text{C NMR}$ δ 172.7, 151.0 (ddd, $J = 243.8, 8.5, 1.7$), 146.9 (ddd, $J = 241.6, 12.2, 3.2$), 144.7 (ddd, $J = 242.3, 13.9, 12.4$), 137.4 (ddd, $J = 9.6, 6.1, 2.9$), 107.9 (dd, $J = 20.7, 4.1$), 106.4 (dd, $J = 26.6, 21.5$), 55.9, 53.6 (2C), 51.3 (d, 2C, $J = 3.0$), 45.4 (2C), 40.1, 38.0 (2C), 36.9, 24.7 (2C); ESI-MS m/z 410 (MH⁺). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid. The solid was recrystallized from hot methanol/chloroform (4:1) (with hexane added to cloudiness), giving white flakes (0.43 g, 0.96 mmol, 22%) (mp 234–236.5 °C).

8-((1*R*)-2-[4-(2,4,5-Trifluorophenyl)piperazin-1-yl]-1-methylethyl)-8-azaspiro[4.5]decane-7,9-dione (12). A mixture of 1-(2,4,5-trifluorophenyl)piperazine (105 mg, 0.49 mmol) and (*R*)-8-(2-chloro-1-methylethyl)-8-azaspiro[4.5]decane-7,9-dione (105 mg, 0.43 mmol) was heated with stirring at 160 °C for 5 h. The residue was dissolved in methanol, transferred to a preparative thin layer chromatographic plate (silica gel), and eluted with ethyl acetate/hexane (1:1). A band at $R_f = 0.8$ was removed and rinsed with chloroform/methanol (4:1). The

solvent was removed, giving the title compound as a pale-yellow oil (86.2 mg, 0.20 mmol, 47%, 100% ee). $^1\text{H NMR}$ δ 6.89 (ddd, 1H, $J = 11.7, 10.2, 7.5$), 6.71 (dt, 1H, $J = 12.0, 8.1$), 5.08–4.96 (m, 1H), 3.14 (dd, 1H, $J = 12.6, 10.5$), 2.92 (t, 4H, $J = 4.7$), 2.73–2.66 (m, 2H), 2.58 (s, 4H), 2.51–2.44 (m, 2H), 2.36 (dd, 1H, $J = 12.6, 5.4$), 1.75–1.68 (m, 4H), 1.57–1.50 (m, 4H), 1.34 (d, 3H, $J = 6.9$); ESI-MS m/z 424 (MH^+). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (mp 231–235 °C).

8-[(1S)-2-[4-(2,4,5-Trifluorophenyl)piperazin-1-yl]-1-methylethyl]-8-azaspiro[4.5]decane-7,9-dione (13). **13** was prepared in a manner analogous to that described above for **11**, giving the title compound as a pale-yellow oil (18.1 mg, 0.043 mmol, 52%, 98% ee). $^1\text{H NMR}$ δ 6.89 (ddd, 1H, $J = 11.7, 10.2, 7.5$), 6.71 (dt, 1H, $J = 12.0, 8.1$), 5.08–4.96 (m, 1H), 3.14 (dd, 1H, $J = 12.6, 10.5$), 2.92 (t, 4H, $J = 4.7$), 2.73–2.66 (m, 2H), 2.58 (s, 4H), 2.51–2.44 (m, 2H), 2.36 (dd, 1H, $J = 12.6, 5.4$), 1.75–1.68 (m, 4H), 1.57–1.50 (m, 4H), 1.34 (d, 3H, $J = 6.9$); ESI-MS m/z 424 (MH^+). The title compound was dissolved in ether and precipitated by addition of 1 N HCl in ether, giving a white solid (mp 231–235 °C).

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Supporting Information Available: Results from elemental analysis. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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